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ACIDIC COMPOUNDS IN PATCHOULI OIL

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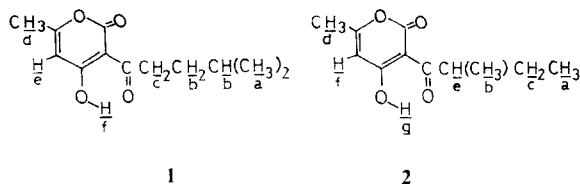
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Key Word Index—*Pogostemon cablin*; Labiatae, patchouli oil, 3 - (4 - Methylpentanoyl -)3,4 - dihydro - 6 - methyl - 1,2 - pyran - 2,4 - dione, 3 - (2 - Methylbutyryl -)3,4 - dihydro - 6 - methyl - 1,2 - pyran - 2,4 - dione

Indonesian patchouli oil obtained from *Pogostemon cablin* Benth. by steam distillation gave a yellowish oil on treatment with 1N Na₂CO₃. The yellowish oil was separated by means of column chromatography and the 5% Et₂O-*n*-hexane fraction was examined. The gas chromatogram of this fraction has 2 main peaks at RT 12.4 min. (peak 1) and 7.6 min. (peak 2), the proportions of peak area being 85 and 12% respectively. Recrystallization with absolute ethanol gave needles of **1** (mp 34.5–35.5°), with the same retention time as peak 1. Semicarbazone mp 184.8–185.3° (dec.). On boiling with aq 0.5N NaOH, **1** yielded isoamylmethyl ketone as the main neutral product. IR spectrum of **1**: 3110, 1743, 1723, 1643, 1609, 1565, and 998 cm⁻¹. MS; *m/e* 224 (M⁺, 4%), 209 (M⁺-Me, 3%), 181 (M⁺-C₃H₇, 40%), 168 (M⁺-56, 100%), 153 (M⁺-71, 81%), 85 (O≡C-C≡C-C≡C-O, 28%), and 43 (C₃H₇, 44%). Elemental analysis of **1** Found; C 64.16%, H 7.25%, N 0.0%, calc. for C₁₂H₁₆O₄; C 64.28%, H 7.20%, O 28.54%. NMR (δ_{ppm}^{CCl₄}); (a) 0.96 (*d*, 6H), (b) 1.55 (*m*, 3H), (c) 2.95 (*t*, 2H), (d) 2.23 (*s*, 3H), (e) 5.80 (*s*, 1H), and (f) 16.87 (*s*, 1H). From the above data structure **1** was deduced for this substance. The compound of peak(2) was isolated in the pure state by means of preparative GLC. IR of **2**: 3110, 1743, 1724, 1642, 1608, 1548, 1462, and 998 cm⁻¹. MS; *m/e* 210 (M⁺, 58%), 168 (M⁺-56, 15%), 153 (M⁺-C₄H₉, 100%), 126 (M⁺-O≡C-C₄H₉, 26%), 85 (O≡C-C≡C-C≡C-O, 34%),

43 (C₃H₇, 29%), and 29 (C₂H₅, 10%). NMR (δ_{ppm}^{CCl₄}); (a) 0.93 (*t*, 3H), (b) 1.12 (*d*, 3H), (c) 1.6 (*m*, 2H), (d) 2.25 (*s*, 3H), (e) 3.77 (*m*, 1H), (f) 5.81 (*s*, 1H), and (g) 17.18 (*s*, 1H). The structure assumed for peak(2) from these data is **2**. These two compounds were synthesized by the condensation of corresponding acid chloride with triacetic acid lactone in the presence of H₂SO₄[1]. IR, MS, and NMR of the synthetic specimens were completely in agreement with those of the natural compounds. The formal names for **1** and **2** are 3 - (4 - methylpentanoyl -)3,4 - dihydro - 6 - methyl - 1,2 - pyran - 2,4 - dione and 3 - (2 - methylbutyryl -)3,4 - dihydro - 6 - methyl - 1,2 - pyran - 2,4 - dione respectively. That these structures are in tautomerism with the corresponding enol form, 3 - (4 - methylpentanoyl -)4 - hydroxyl - 6 - methyl - α - pyrone and 3 - (2 - methylbutyryl -)4 - hydroxy - 6 - methyl - α - pyrone is indicated by the NMR spectra.



The presence of two compounds in the original oil was confirmed by the direct GLC of the oil.

EXPERIMENTAL

Analytical methods For column chromatography Kieselgel-G was used with *n*-hexane, 5%-Et₂O-*n*-hexane, 10%-Et₂O-*n*-hexane, 50%-Et₂O-*n*-hexane and Et₂O as elut-

ing solvents. For analytical and preparative GLC, Shimadzu GC-4AP was used with OV-1 as the stationary phase at 180°. NMR spectra were obtained using TMS as the internal standard. IR spectra were obtained in liquid film.

Triacetic acid lactone. A soln of dehydroacetic acid (20 g) in 90% H₂SO₄ (100 g) was heated in an oil bath at 140° for 1 min, the reaction mixture was poured on crushed ice immediately. Resulting crystals were filtered and recrystallized from diluted ethanol. MP 187°. Yield 6.9 g.

4-Methylpentanoyl chloride. 4-Methylpentanoic acid (23.2 g) was added dropwise into SOCl₂ (28.6 g) under refluxing during the course of 30 min and the reaction mixture was refluxed for 1 hr. It was distilled directly and the fraction of bp 119–131° was collected. Yield 17.4 g.

2-Methylbutyryl chloride. DL-2-Methylbutyric acid (16.7 g) was treated by the same way as above bp 106–119°C. Yield 14.8 g

3 - (4 - Methylpentanoyl -)3,4 - dihydro - 6 - methyl - 1,2 - pyran - 2,4 - dione. Triacetic acid lactone (3.1 g) was added to 4-methylpentanoyl chloride (9.7 g) previously mixed with 0.5 ml of H₂SO₄ and cooled with ice H₂O below 10°. The mixture was gradually heated and kept at 85–90° for 30 min. After generation of HCl gas ceased the reaction mixture was poured onto crushed ice. The resulting oil was extracted with *n*-hexane, washed with H₂O, dried and evaporated. The oil obtained was purified by means of column chromatography using Kieselgel-G and then alumina. Yield 0.65 g

3 - (2 - Methylbutyryl -)3,4 - dihydro - 6 - methyl - 1,2 - pyran - 2,4 - dione. DL-2-Methylbutyryl chloride was treated in the same way as above. Yield 0.2 g.

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A NEW DITERPENOID ACETATE FROM *SIDERITIS REVERCHONII**

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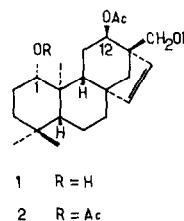
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Key Word Index—*Sideritis reverchonii*; Labiatae; *ent* - 15 - beyerene, *ent* - 16S - atis - 13 - ene and *ent* - 8,5 - friedopimar - 5 - ene derivatives.

Plant. *Sideritis reverchonii* Wk. **Source.** Reduced areas in the S. of the Iberian Peninsula (Cerro de San Antón, near Málaga). **Previous work.** None. **Present work.** From the aerial parts of the species quoted above have been isolated eight diterpenoids previously described: *ent* - 8,5 - friedopimar - 5 - ene - 15S,16 - diol (lagascol) [1], *ent* - 8,5 - friedopimar - 5 - ene - 11β,15S,16 - triol (lagascatriol) [2], *ent* - 15 - beyerene - 12α,17 - diol (tobarrol) [1], *ent* - 15 - beyerene - 7α,17 - diol (benuol) [1], *ent* - 15 - beyerene - 1β,12α,17 - triol (jativatriol) [3], *ent* - 15 - beyerene - 7α,12α,17 - triol (conchitriol) [3], *ent* - 16R - atis - 13 - ene - 16,17 - diol (serradiol) [1] and *ent* - 16R - atis - 13 - ene - 1β,16,17 - triol (sideritol) [4], plus a new compound 1 (C₂₂H₃₄O₄), which when subjected to the action of Ac₂O in pyridine solution affords 2, identical (physical and spec-

troscopic data) with the triacetate of jativatriol. The NMR spectrum of 1 shows a narrow multiplet at δ 5.18 assigned to the equatorial proton geminal to the C-12 acetoxy function [3].



The new diterpenoid (1) is thus 12-acetyl jativatriol (*ent* - 12α - acetoxy - 15 - beyerene - 1β,17 - diol).

EXPERIMENTAL

For general details on extraction and separation of diterpenes from *Sideritis*, see part 17[1] of this series; for most experimental details, see part 13[3]. The previously described compounds have been characterized by their physical and spectroscopic (IR and NMR) data and by

* Part 26 in the series 'Studies on diterpenes from *Sideritis* genus'. For part 25 see von Carstenn-Lichterfelde, C., Rodríguez, B. and Valverde, S. (1975) *Experientia* 31, 757.